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ABSTRACT: A straightforward, cheapest spectrophotometric technique has been developed to measure both Metoprolol (MET) and Amlodipine (AML) simultaneously in dosage and bulk forms. The technique allows for the collection of data at various wavelengths to produce linear graphs, which is used to calculate concentrations. The absorbance values of the two analytes were linear with the concentration at the wavelengths taken at 10 nm interval over the range of 230 -300 nm. The accuracy and the precision of the new approach demonstrated were very good (RSD < 2%). Developed method was validated in terms of linearity, range, precision, accuracy. The suitability of all methods for quantitative determination of these compounds was confirmed by validation in accordance with the requirements of ICH Guidelines.

KEYWORDS: Spectrophotometry, Multi-wavelength, Regression analysis, Metoprolol, Amlodipine

Introduction^{[1][2][3]}

There are variety of intravenous and oral formulations available in market for the treatment of hypertension, angina, heart failure, myocardial infarction, and atrial fibrillation/flutter. The choice of beta-blockers for the treatment of the aforementioned disorders is a topic of debate. Metoprolol is a cardio selective beta-1-adrenergic receptor inhibitor that competitively blocks beta1 receptors with minimal or no effects on beta-2 receptors and decreases cardiac output by negative inotropic and chronotropic effects. Administration of metoprolol to normal subjects results in a reduction in heart rate and cardiac output; this appears to be related to the dose and concentration of the drug.

Amlodipine inhibits the initial calcium influx by blocking the voltage-dependent L-type calcium channels. Vasodilation is the outcome of enhanced smooth muscle relaxation and decreased vascular smooth muscle contractility caused by reduced intracellular calcium. Amlodipine has also been shown to enhance vascular endothelial function in hypertensive

individuals. Amlodipine lowers blood pressure by relaxing smooth muscles and widening blood vessels.

They work together to improve the heart's efficiency in pumping blood throughout the body.

Literature search revealed that different methods have been developed for the simultaneous determination of MET and AML as mixture as well as in dosage form; absorbance ratio technique and difference spectrophotometric method, derivative spectrophotometric, simultaneous estimation methods as well as chemo metric spectroscopic method. The observed resolution between the individual absorption peaks of the two sample components is the only factor that influences the analytical approach chosen for the simultaneous spectrophotometric measurement of these components. In cases where the absorption peaks can be well resolved, this determination is straightforward. However, chemometric procedures become necessary if there is partial or complete overlap between the signals of the various components. These methods necessitate full-spectrum data and highly specialized software for the processing of the spectral data. Therefore, this research work's goal was to create a straightforward, affordable, and precise spectrophotometric technique based on multi-wavelength regression analysis for the simultaneous measurement of amlodipine and metoprolol in tablet and bulk dose forms.

Theoretical Consideration:^[18]

Multi-Wavelength Linear Regression Analysis; enables determination of the composition of a binary mixture with overlapping spectra without determining molar absorptivities or complicated mathematics. The method is very simple it requires only three measurements, the absorbance of a standard solution for each component, and the unknown mixture itself. Assuming additivity, the absorbance of a mixture is the sum of the absorbancies of its components. If we have a mixture consisting of two components, X and Y, with an unknown concentration of Cx and Cy, then the absorbance of the unknown mixture,

However, the absorbencies of standard solutions of the same substances will follow the same Beer's law relationship and have the same molar absorbance, \mathcal{E} , and one centimetre path length, b, as the unknown solutions under the same conditions.

A	A	A
Cmix = Cx+Cy	Csx (standard X)	Csy (standard Y)
$A_{mix} = A_x + A_y$	$Asx = \epsilon_x bCs_x$	Asy = € _y bCs _y
$A_{mix} = \pounds_x bC_x + \pounds_y bCy2$	$ \in_{\mathbf{x}} = \mathbf{Asx}/\mathbf{bCs_x} \dots \mathbf{a}_3 $	$\epsilon_y = Asy/bCs_{y4}$
Diving eq ⁿ 2 by Asx	Diving equation 3 b	y 4, $\frac{\notin y}{\notin x} = \frac{Asy}{Asx} \cdot \frac{Csx}{Csy}$ 5
$\frac{Amix}{Asx} = \frac{\notin xbCx}{Asx} + \frac{\notin ybCy}{Asx}$ [substituting	g Asx from eq ⁿ 1]	
$\frac{\text{Amix}}{\text{Asx}} = \frac{\text{\&xbCx}}{\text{\&xbCsx}} + \frac{\text{\&ybCy}}{\text{\&xbCsx}}$		
$\frac{Amix}{Asx} = \frac{Cx}{Csx} + \frac{\notin y}{\notin x} \cdot \frac{Cy}{Csx}$ [substituting	$g \frac{\notin y}{\notin x}$ from eq ⁿ 5]	
$\frac{Amix}{Asx} = \frac{Cx}{Csx} + \frac{Asy}{Asx} \cdot \frac{Csx}{Csy} \cdot \frac{Cy}{Csx}$		
$\frac{Amix}{Asx} = \frac{Cx}{Csx} + \frac{Asy}{Asx} \cdot \frac{Cy}{Csy}$		
Comparing eq ⁿ 6 with regression equa	tion, $Y = \frac{Amix}{Asx}$ slope	$=\frac{Cy}{Csy}$
	$X = \frac{Asy}{Asx}$ inter	$cept = \frac{Cx}{Csx}$

From eqⁿ 7 and eqⁿ 8, Cx = Intercept * Csx and Cy = Slope * Csy

 $\frac{\text{Amix}}{\text{Asx}} = \frac{\text{Cx}}{\text{Csx}} + \frac{\text{Asy}}{\text{Asx}} \cdot \frac{\text{Cy}}{\text{Csy}}$

Comparing eqⁿ 6 with regression equation, $Y = \frac{Amix}{Asx}$ slope = $\frac{Cy}{Csy}$

$$X = \frac{Asy}{Asx} \qquad \text{intercept} = \frac{Cx}{Csx}$$

From $eq^n 7$ and $eq^n 8$, Cx = Intercept * Csx and Cy = Slope * Csy

That is, the concentration of the unknown component Y (Cy) in the mixture, equals the slope times the concentration of the standard solution for component Y. Likewise, the concentration of the unknown component X (Cx) in the mixture equals the product of the intercept times the concentration of the standard solution for component X.

Materials and Methods

Instruments

Spectroscopic analysis was carried out on a UV- VISIBLE 1900 (Shimadzu) double beam UV-Visible spectrophotometer with software of UV probe version 2.70. The absorption spectra were recorded over the wavelength range of 200-400nm, against solvent blank in quartz cuvettes with 1cm diameter. Data analysis was performed using Microsoft Excel Spreadsheet 2016.

Materials

Active Pharmaceutical Ingredient (API) Metoprolol and Amlodipine was received as gift sample from Torrent Pharma. Marketed formulation MPROL AM 25/5 tablet was purchased from local medical store.

Standards and solutions:

Preparation of standard stock solution of Metoprolol:

Weigh 25mg of Metoprolol and dissolved in 25 ml of methanol. Make up the volume up to mark with methanol to get a solution containing 1000ppm Metoprolol. Now take 2.5ml of standard stock solution and dilute up to 25ml with methanol to get a solution containing 100ppm Metoprolol. Take 0.5ml, 1ml, 1.5ml, 2ml, 2.5ml from 100ppm stock solution of Metoprolol and transfer to five different volumetric flask of 10ml to get 5,10,15,20,25 ppm of Metoprolol solution respectively.

Preparation of standard stock solution of Amlodipine by Standard Addition Method:

Weigh 25mg of Amlodipine and dissolved in 25ml of methanol. Make up the volume up to mark with methanol to get a solution containing 1000ppm Amlodipine. Now take 2.5ml of standard stock solution and dilute up to 25ml with methanol to get a solution containing 100ppm Amlodipine. Take 0.1ml, 0.2ml, 0.3ml, 0.4ml, 0.5ml from 100ppm stock solution of Amlodipine and transfer to five different volumetric flask, add additional 1ml of standard stock solution(100ppm) of Amlodipine to each flask and then make up the volume up to 10 ml with methanol to get 11,12,13,14,15 ppm of Amlodipine solution respectively.

Preparation of sample preparation

Twenty tablets were accurately weighed and crushed to a fine powder; the weight of powder equivalent to one tablet was transferred into a 25mL volumetric flask, dissolved with the methanol to get a solution containing 1000ppm Metoprolol and 200ppm Amlodipine. Take 0.05, 0.1, 0.15, 0.2 and 0.25 ml of above solution in five different volumetric flasks of 10ml and make up the volume up to the mark with methanol to get 5,10,15,20,25 ppm

metoprolol and 11,12,13,14,15 Amlodipine For the application of the method, the ratio of the absorbance of mixture and sample absorbance of amlodipine analytical standard solution at each wavelength was calculated and plotted against the ratio of absorbance values obtained for amlodipine analytical standard solution to the same Metoprolol standard solution used in the first instance. The slope and intercept of the straight

line obtained were then used to calculate the concentration of each analyte in the mixture or the sample.



solution respectively.

Figure 1: Absorption spectra of (a) Metoprolol 10 μ g/ml and (b) Amlodipine 12 μ g/ml.

Methods

Spectral characteristics:

The individual spectra of the analytes were obtained by scanning their corresponding working standard solutions over the wavelength range of 230-300 nm.

Linearity

The absorbances of the linearity standard solutions were read at 10 nm intervals over the wavelength range of 230-300 nm

General procedure :

The absorbance values of the two working standards, synthetic mixtures and samples were read at10 nm intervals over the wavelength range of 230-280 nm. The concentrations of MET and AML in the synthetic mixtures and the samples were calculated according to the MLRA principle using the slopes and intercepts of the straight lines obtained.

Results and discussion

Spectral characteristics

The individual spectra of the analytes showed extensive overlapping over the wavelength range of 230-300 nm (Fig 1);hence resolutions of their photometric signals by classical spectrophotmetric methods are not possible.

Linearity

Application of MLRA method requires obeying Beer's law at each wavelength over the selected wavelength range. The two analytes showed linear relation between the concentration and absorbance over the entire wavelengths range investigated ($r_2 < 0.99$).

PARAME TEP	WAVELENGTH						
ILN	220	230	240	250	260	270	
b	0.0389	0.0241	0.0027	0.0027	0.0036	0.0055	
а	-0.0124	-0.0122	-0.0045	-0.0026	-0.0026	-0.0026	
r ²	0.9984	0.9982	0.9935	0.9947	0.9947	0.9977	
Sb	0.0005	0.0009	0.0007	0.0009	0.0008	0.0009	
Sa	0.0019	0.0028	0.0007	0.0009	0.0003	0.0008	
LOD	0.1612	0.3834	0.8556	1.1000	0.2750	0.4800	
LOQ	0.4884	1.1618	2.5926	3.3333	0.8333	1.4545	

Table 1 : Linearity data of Metoprolol

b slope, a intercept, r^2 correlation coefficient, S_b Standard deviation of slope, S_a Standard deviation of Intercept, LOD limit of detection, LOQ limit of Quantitation.

Table 2 : Linearity data of Amlodipine

PARAME	WAVELENGTH						
TER	220	230	240	250	260	270	
b	0.4545	0.0491	0.0352	0.0156	0.0069	0.0058	
a	0.0264	-0.1237	0.0302	0.0792	0.0409	-0.0096	
r ²	0.9905	0.9927	0.994	0.9941	0.9853	0.9929	
Sb	0.002	0.0035	0.0044	0.0007	0.0005	0.0003	
Sa	0.0008	0.0009	0.0007	0.0009	0.0003	0.0004	
LOD	0.1454	0.2352	0.4125	0.1481	0.2391	0.1707	
LOQ	0.4405	0.7128	1.2500	0.4487	0.7246	0.5172	

b slope, a intercept, r^2 correlation coefficient, S_b Standard deviation of slope, S_a Standard deviation of Intercept, LOD limit of detection, LOQ limit of Quantitation.

Determination of synthetic mixtures (Accuracy)

The accuracy of the proposed method was tested by analyzing laboratory prepared synthetic mixtures containing different concentration ratios of METOPROLOL and AMLODIPINE at four different level i.e. at 0%, 80%, 100% and 120%. The results obtained indicated good accuracy of the method supporting the suitability of its application for the analysis of the two drugs in Tablet

SAMPLE	METOPROLOL AMLODIPINE					
	Theoretical Yield	Practical Yield	% Recovery	Theoretical Yield	Practical Yield	% Recovery
M1	10	10.1	101	12	12.24	102
M2	18	18.52	101.88	13.6	13.8	101.47
M3	20	20.2	101	14	14 14.01	
M4	22	22.95	99.77	14.4	14.5	100.28
Average		100.912	Average		100.955	
SD		0.8673	SD		0.9302	
% RSD		0.8594	% RSD 0.92		0.92145	

 Table 3: Accuracy data of Metoprolol and Amlodipine

Method precision

a. Repeatability: The standard solution containing 10ppm Metoprolol and 12ppm Amlodipine was prepared six times in a day using same apparatus, instruments and chemical. Then absorbance was noted and %RSD was calculated.

Wavelength	METOPROLOL		AMLODIPINE		
	ABSORBANCE(MEAN)(n=6) \pm SD		ABSORBANCE(MEAN)(n=6) \pm SD		
	10 ppm	%RSD	12 PPM	%RSD	
220	0.373±0.0008	0.2019	0.540±0.0043	0.7934	
230	0.227±0.0008	0.3319	0.463±0.0014	0.3054	
240	0.022±0.0010	0.6625	0.434±0.0005	0.1191	
250	0.023±0.0003	1.1308	0.246±0.0009	0.3636	
260	0.034±0.0006	1.0114	0.114±0.0012	1.0791	
270	0.055±0.0005	1.1499	0.058±0.0008	1.2942	

 Table 4: Repeatability data

b. Intermediate Precision: Six runs of standard 15ppm Metoprolol and 13ppm Amlodipine was prepared for intermediate precision. Absorbance was noted and %RSD was calculated.

Table 5: Metoproloi and amlodipine intermediate precision data							
Wavelength	METOPROLOL ABSORBANCE(MEAN)(n=3) ± SD						
	10 ppm	%RSD	15 ppm	%RSD	20 ppm	%RSD	
220	0.373±	0.2268	0.5660±	0.1767	0.7477±	0.0772	
	0.0010		0.0010		0.0006		
230	0.226±	0.2547	0.3437±	0.1680	0.4603±	0.2508	
	0.0006		0.0006		0.0012		
240	$0.021\pm$	0.6891	0.3100±	0.1854	$0.0443 \pm$	1.3023	
	0.0010		0.0100		0.0007		
250	$0.022 \pm$	1.2642	0.0320±	1.7856	0.0467±	1.2372	
	0.0006		0.0010		0.0006		
260	0.03±	0.8872	0.0467±	1.1395	0.0693±	0.8327	
	0.0006		0.0058		0.0010		
270	$0.055\pm$	1.0434	0.0820±	1.2195	0.113±	0.5186	
	0.0010		0.0010		0.0006		
Wavelength		AMLODIP	INE ABSORBA	ANCE(MEAN)($(n=3) \pm SD$		
	12 ppm	%RSD	13 ppm	%RSD	14 ppm	%RSD	
220	0.539±	0.0965	0.601±	0.2543	0.653±	0.2338	
	0.0015		0.0015		0.0015		
230	$0.462 \pm$	0.2165	0.551±	0.1815	$0.555\pm$	0.1802	
	0.0010		0.0010		0.0010		
240	0.434±	0.1331	0.471±	0.2123	0.514±	0.1946	
	0.0010		0.0010		0.0010		
250	$0.246 \pm$	0.4065	0.272±	0.9727	0.338±	0.2959	
	0.0026		0.0026		0.0010		
260	0.113±	1.3478	0.139±	0.7194	0.136±	1.1259	
	0.0010		0.0010		0.0015		
		L	+		+		
270	$0.059\pm$	1.6949	$0.063 \pm$	1.5873	$0.07 \pm$	1.4286	

Table 5: Metoprolol and	amlodipine intermediate	precision data
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Analysis of commercial sample:

Twenty tablets were accurately weighed and crushed to a fine powder; the weight of powder equivalent to one tablet was transferred into a 25mL volumetric flask, dissolved with the methanol to get a solution containing 1000ppm Metoprolol and 200ppm Amlodipine. Now take 0.05, 0.1, 0.15, 0.2 and 0.25 ml of above solution in five different volumetric flasks and add 1 ml of standard amlodipine (10 ppm) and make up the volume up to the mark with methanol to get 5,10,15,20,25 ppm metoprolol and 11,12,13,14,15 ppm amlodipine solution respectively. **Table 6: Sample Assay Result**

SR.	. Concentration		Regression	Intercept	Slope	Concentration of	Concentration
NO.	. (ppm)		Equation	(c)	(m)	Metoprolol	of Amlodipine
	METO	AMLO	(y=mx+c)			(ppm)	(ppm)
	(Csx)	(Csv)				$[C_x =$	[C y =
						Intercept*C _{sx}]	Slope*C _{sy}]
1	5	11	Y = 0.9988x	1.2667	0.9788	5.333	10.986
			+ 1.0667				
2	10	12	Y = 1.0242x	1.0095	1.0242	10.095	12.290
			+ 1.0095				
3	15	13	y = 1.009x +	0.998	1.009	14.97	13.117
			0.998				
4	20	14	y = 1.0344x	1.0305	1.0344	20.61	14.481
			+ 1.0305				
5	25	15	y = 1.031x +	1.0095	1.031	25.237	15.465
			1.0095				

Conclusion:

Multi-wavelength Linear Regression Analysis is a straightforward procedure allowing the accurate resolution of binary mixtures of compounds with overlapped spectra without prior separation, the method is very simple requires only measurement of absorbance values of the two standards and the sample. The concentrations of the two analytes are calculated using easy to understand simple mathematics. The cost effectiveness in term of time and money renders the method as suitable alternative to other expensive techniques e.g., chromatographic methods for the analysis of binary mixtures of compounds with overlapped spectra in laboratories and countries where such sophisticated equipment are not affordable. The accuracy and simplicity of the method suggest it suitability in cases where quick results are demanded e.g., as in-process analysis procedure during blend analysis in industrial setups.

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